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Constructing the 2-(thiobenzyl)ethyl carbamate linker via thiyl radical addition

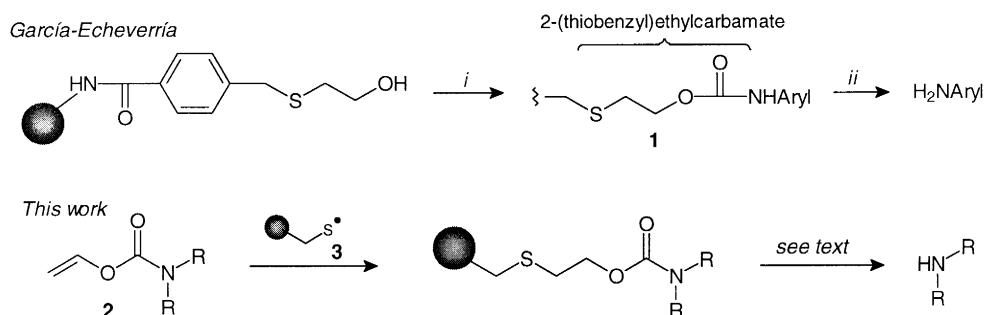
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Abstract

N-Vinylloxycarbonyl derivatives of secondary amines undergo efficient addition of thiyl radical and, using Merrifield SH resin, this reaction provides a new entry to resin-bound 2-(thiobenzyl)ethyl carbamates. © 2000 Elsevier Science Ltd. All rights reserved.

A variety of methods exist for linking amines to resin supports for use in solid phase synthesis.^{1–5} Many of these methods are based on protocols developed in the area of peptide chemistry, and limitations of the linkers used become apparent with less traditional amine substrates. As a consequence, alternative linkers have been developed and recently García-Echeverría described the use of 2-(thiobenzyl)ethyl carbamate **1** as a base labile traceless linker.⁶ The carbamate unit is established via alcohol addition to an isocyanate, and cleavage of the desired primary amine from **1** is achieved using an oxidation/elimination sequence (Scheme 1).⁷

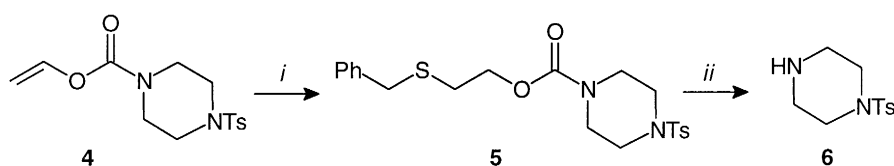


Scheme 1. Reagents: (i) ArylNCO; (ii) (following chemical modification of the aryl moiety) 4 equiv. *m*CPBA, then 10% NH₄OH in 2,2,2-trifluoroethanol

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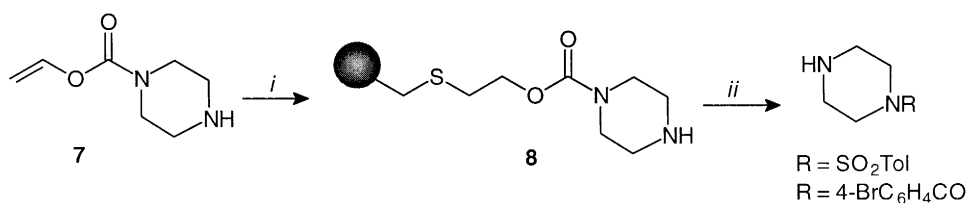
In this paper we describe an alternative method for constructing the 2-(thiobenzyl)ethyl carbamate unit based on the ability of an *N*-vinylloxycarbonyl (VOC) derivative **2** to undergo addition of a benzylic thiyl radical **3**.⁸ This methodology provides a method for attaching secondary amines (and thereby complements the isocyanate method outlined in Scheme 1) and serves to extend the synthetic utility of this flexible traceless linker. While we have used García-Echeverría's oxidation/elimination sequence to release the product amine, we also suggest an improvement to this original protocol which has given higher yields of final products, together with an alternative method for cleavage of the linker based on *S*-alkylation.

The underlying efficiency of the thiyl radical addition step was evaluated in solution, and the process has also been applied to solid phase using simple piperazines as model substrates. Under solution conditions, radical addition of benzylmercaptan to *N*-VOC **4** (Ts=SO₂Tol) was best accomplished under strictly O₂-free conditions to give **5** in 87% isolated yield. Using a modification of García-Echeverría's original conditions, oxidation of **5** followed by treatment of the corresponding sulfone with either DBU or *N,N,N',N'*-tetramethylguanidine in CH₂Cl₂ gave amine **6** in 81% yield (Scheme 2).⁹



Scheme 2. Reagents: (i) PhCH₂SH, AIBN, DMF, 80°C (87%); (ii) *m*CPBA (5 equiv.), CH₂Cl₂, then DBU, CH₂Cl₂, rt, 2 h (81%)

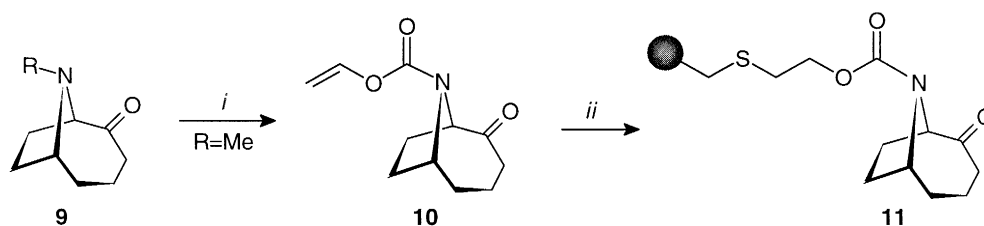
For the application of this methodology to solid phase we have focused on the Merrifield SH resin as the source of thiyl radicals, although a number of alternative thiol resins are commercially available. Merrifield SH resin¹⁴ must be primed before use by reduction with 1,4-dithiothreitol (DTT) to cleave disulfides and release reactive thiol sites. The monoacylated *N*-VOC piperazine **7** was then incorporated onto the resin under radical conditions (AIBN, DMF, 80°C) and, based on microanalytical data, the thiyl addition process proceeded with high efficiency (Scheme 3). The resulting resin **8** was then derivatised with either toluenesulfonyl chloride or 4-bromobenzoyl bromide. The corresponding monosubstituted piperazine products were released using the modified oxidation/elimination procedure in 47% and 50% overall yield (>90% purity based on HPLC analysis using an internal standard and authentic materials), respectively.¹⁴



Scheme 3. Reagents: (i) Merrifield SH resin, AIBN, DMF, 80°C; (ii) TolSO₂Cl/py or 4-BrC₆H₄COBr/py; (iii) *m*CPBA (5 equiv.), CH₂Cl₂, then DBU (47% and 50% overall yield from **7** based on a loading level of 0.31 mmol gm⁻¹ of thiol)

Any synthetic process is enhanced when a choice of more than one preparative protocol is available, but an ability to construct the 2-(thiobenzyl)ethyl carbamate by exploiting a pre-existing carbamate moiety does have other advantages particularly when substrates containing amine-sensitive functionality, for example aminoketones, are involved (Scheme 4). We have already employed azabicycloketones, such as **9**, as an entry to novel and potent nicotinic agonists,¹⁵ and while the *N*-methyl variants (**9**, R=Me) are readily accessible, the corresponding free secondary amines (**9**, R=H) are prone to rapid degradation.

Vinyl chloroformate serves to effect a facile *N*-demethylation of (**9**, R=Me) and concomitant *N*-protection to give the *N*-VOC derivative **10** in excellent yield, which underwent thiyl radical addition (using the primed Merrifield SH resin under the conditions described above) to give **11**.



Scheme 4. Reagents: (i) vinyl chloroformate, K₂CO₃, CH₂Cl₂ (96%); (ii) Merrifield SH resin, AIBN, DMF, 80°C

To demonstrate the basic feasibility of this strategy, we have carried out a simple reduction of the resin bound aminoketone **11** (using NaBH₄) and released the corresponding alcohol (as a mixture of diastereoisomers) using the oxidation/elimination procedure.¹⁶ Studies are now underway to assess the compatibility of **11** (and related derivatives) towards a wider range of synthetically useful transformations, including metal-mediated cross coupling reactions, to determine the broader utility of the sulfur-containing 2-(thiobenzyl)ethyl carbamate linker.

In summary, vinyloxycarbonyl derivatives of secondary amines undergo efficient thiyl radical addition which provides an alternative and complementary method for constructing the traceless 2-(thiobenzyl)ethyl carbamate linker.

Acknowledgements

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- An alternative method for cleaving this linker has been examined which exploits the high nucleophilic character of the sulfide moiety. Selective *S*-alkylation¹⁰ of **5** using triethyloxonium tetrafluoroborate, followed by methanolic workup gave amine **6** in 50% overall yield. This procedure was successfully applied to the corresponding resin bound substrate, and reaction of **8** with TsCl followed by Et₃OBF₄ and MeOH workup gave **6** in an unoptimised 23% overall yield. It is likely that this process involves C–S bond cleavage via neighboring group participation of the carbonyl oxygen as the mechanism for carbamate cleavage,^{11–13} and this may offer possible synthetic advantage over the oxidation/base-mediated elimination protocol under appropriate circumstances.
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14. Activation of resin. We prepared Merrifield SH resin (Fréchet, J. M. J.; de Smet, M. D.; Farrall, M. J. *Polymer* **1979**, *20*, 675–680), from Merrifield LL 200–400 mesh chlorine resin (containing 0.74 mol/gm of chloride). A good conversion was observed (Anal. calcd for 100% conversion: S, 2.37; Cl, 0%. Found: S, 2.36; Cl, 0.36%). The Merrifield SH resin must be reduced prior to use according to the following procedure, which is based on 1 g of resin and the procedure must be carried out under nitrogen. The resin was prewashed three times using DMF (10 cm³), then allowed to swell in DMF (10 cm³) for 10–15 minutes. DTT (771 mg) and TRIS–EDTA–HCl buffer pH 7.5 (Ellman-Test-Buffer) (2 cm³) were added and the mixture was kept at room temperature overnight. After this time, acetic acid (5 cm³) was added and the resin was filtered, washed several times with DMF, ethanol or with the solvent to be used in subsequent reactions. In ‘reduced’ resin, the SH moiety is observed at 2580 cm⁻³, which is absent in the resin prior to reduction with dithiothreitol. Based on treating reduced resin with FmocOSu (to trap accessible -SH groups), followed by Fmoc cleavage, we estimate the loading of SH (as opposed to other S-functionalities) to be 0.31 mmol/g. The yields shown in Scheme 3 are based on this level of thiol loading, but we have assumed that all subsequent reactions (thiyl radical addition, derivatisation of **8**, and cleavage) proceed quantitatively. The yields reported must, therefore, be regarded as a lower limit. García-Echeverría⁶ originally used 10% NH₄OH in 2,2,2-trifluoroethanol as the reagent for achieving cleavage of the resin-bound sulfone, although no chemical yields were actually reported. Using his conditions, we only observed a 24% yield of piperazine **6**, as compared to 47% using the modified procedure.
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